HORIBA Explore the future



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Modern Particle Characterization Techniques Introduction

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Why Characterize Particles?

Particle physical properties material behavior and processing across a number of industries

Industry		Industry	
Ceramic		Construction	
Oil/rubber		Chemical	
Battery		Pharmaceutical	
Electricity		Food/Drink	
Automobile		Paper/Pulp	
Mining		Ink/Toner	



How will the data be used?

Physical Property (Scratching Power) (Rate of Surface Removal) (Rate of Dissolution) (Rate of Aggregation) (Dispersion Stability)

(Hiding Power)



Particle Size Parameter

 $(D_{50} \text{ Value})$ $(\% > 1.0 \ \mu\text{m})$ $(\% > 0.5 \ \mu\text{m})$ (Overall Distribution)



Particle Characterization Methods

Particle Sizing
Particle Counting
Surface Area
Porosimetry
Zeta Potential
Particle Shape
Many More



Particle Sizing Techniques

Sedimentation

Coulter Counter

Laser Light Scattering

Dynamic Light Scattering

Image Analysis

Nano Particle Tracking

Many More



Size Range by Technique (µm)





Electrical Zone Sensing

- Coulter Principle
 - Based on change in conductivity of aperture as particle traverses.
 - Requires conducting liquid.
 - Directly measures particle volume and counts.
 - High resolution
 - Used for blood cell counting more than industrial applications







Sedimentation

Stokes Law

$$D = \sqrt{\frac{18\mu V_p}{(\rho_p - \rho_l)g}}$$

- V_{p} = Settling velocity of discrete particle
- g = Gravity constant
- $\rho_p = Density of Particle$
- ρ_{I} = Density of Carrier Fluid
- μ = Viscosity of Carrier Fluid

Note: assumes settling of spherical particle Under-sizes compared to other techniques if non-spherical

Sedimentation of same density material in a viscous medium









Laser diffraction

Silica ~ 30 nm







Laser Diffraction

- •Particle size $0.01 3000 \, \mu m$
- •Converts angular variations in scattered light to
- particle size distribution
- •Quick, repeatable
- •Most common technique
- •Suspensions & powders



Automated Microscopy

Static: Particles fixed on slide, stage moves slide





Dynamic: Particles flow past camera(s)





DLS – Dynamic Light Scattering



time (microseconds)

Nanogold ~ 50 nm



Particle size < 1 nm – 8 μm
Converts intensity variations in scattered light to particle size distribution

- Quick, repeatable
- •Suspensions

	Z-average Diameter, nm
Run 1	50.5
Run 2	51.1
Run 3	49.2
Run 4	51.5
Run 5	49.7
Run 6	50.9
Avg.	50.5
Std. Dev.	0.9
COV	1.7 %





Sedimentation Particle Sizing Approaches

Homogeneous
Line Start
X-ray Irradiation
Light Irradiation
Cell Moves
Source Moves
Gravity Sedimentation
Centrifugal Sedimentation
Many More



BET Surface Area





Particle Analysis Workflow



HORIBA Scientific

All analytical determinations start with sampling (PCA is no exception)

- A representative sample is the key to drawing any scientific conclusion
- This process is seemingly easy, but really complicated
- It is a science of its own so needs some discussion
- It is especially important for broad size distributions
- It is often overlooked by users



Reliability of selected sampling methods using a 60:40 sand mixture

Sampling Technique	Standard Deviation
Cone and Quartering	6.81
Scoop Sampling	5.14
Table Sampling	2.09
Chute Slitting	1.01
Spinning Riffling	0.146
Random Variation	0.075

Allen, T. (1997). Particle Size Measurement Volume 1: Powder Sampling and Particle Size measurement fifth edition., Page 21. Chapman & Hall.



Sampling from Drums



www.samplingsystems.com



Technique: Chute Riffling

Chute splitting allows sample to vibrate down a chute to vanes which separate the mass into two portions. Each portion moves further where they each are divided into two parts, now giving four parts. This is often continued until 8 or 16 portions are obtained.







Sample Dividers



Laboratory sample divider PT 100

- for pourable powders and granules
- feed size up to 10 mm
- division into 6, 8 or 10 representative samples



Laboratory rotary tube sample divider PKZ 1000

- for pourable powders and granules
- feed size up to 10 mm
- various division ratios



Sample splitter RT

- for bulk materials
- feed size up to max. 50 mm
- division into 2 samples



Technique: Sampling from Beaker

Liquid should be in motion vertically and horizontally to ensure good mixing.

Pipette should be about one-third of the way from the bottom when extracting sample.

Alternative: When mixing powders into a slurry: make paste, pipette from paste





Preparation of a Corundum Particle Size Reference Material



Reference: Joint Research Centre (2018) Certification Report – The Certification of Particle Size distribution of Corundum: ERM-FD069, Luxembourg: Publications office of the European Union



Now you have a sample what comes next? (Sample Preparation)

- Most typical method is to match the process which produced the sample
 - High Energy or low energy
- How are others in your industry doing the analysis?
 - This can be a huge time saver
- Who will possibly be viewing my data?
 - Standard preparation may already be specified
- How do you wish to see your sample?
 - Primary particles or how they truly exist
- Possible challenges in preparation
 - What tools are at my disposal and how do they effect the size
- What will I be trying to do with the data?
 - Data Matching, performance prediction, new research, etc.



Sample Preparation tools at your disposal

- All sorts of dispersing media
 - Organics and Inorganics
- All sorts of dispersants and surfactants
 - McCutcheon's Directory has loads of them
- All sorts of mixing devices
 - Magnetic Mixers all the way to Waring Blenders
- All sorts of ultrasonic devices
 - Baths to high energy probes
- Dispersion available during measurement
 - Assure sample stability throughout an analysis





General Dispersion Procedures





Reference: ASTM International (2016) Standard Guide for Liquid dispersion of Metal Powders and Related Compounds for Particle Size Analysis, Designation: B821 – 10, Page 2.



Recommended Dispersion Procedures

Material	Carrier Liquid	Surfactant	Surfactant Concentration	Ultrasonic Treatment		
				Туре	Power Level, W	Time, mir
Chromium carbide	water	none		none		
				Or ^A		
2		T of B		bath	25	5
Copper	water	Iween 21 ^B	3-5 drops	bath	80	1
-erroalloys	isopropyl alcohol	Tween 21 ^D	10 %	bath	80	1
ron/steel	water	Iween 21 ²	3-5 drops	bath	80	1
Manganese sulfide	water	Iween 21 ²	3-5 drops	bath	80	1
Molybdenum	water	sodium hexametaphosphate	0.01 %	probe or	160	3
				bath	80	10
				orA		3.003.0000
				bath	25	5
Vickel	water	Tween 21 ^B	3–5 drops ^C	bath	80	1
Tantalum	water	sodium hexametaphosphate	0.01 %	probe	160	3
				or		
				bath	80	10
Tantalum carbide	water	sodium hexametaphosphate	0.01 %	probe	160	3
				or		
				bath	80	10
				orA		
				bath	25	5
Tungsten	water	sodium hexametaphosphate	0.01 %	probe	160	3
				or		
				bath	80	10
				or		
				bath	25	5
Tungsten carbide	water	sodium hexametaphosphate	0.01 %	probe	160	3
				bath	80	10
				orA	00	10
				bath	25	5

^C Three to five drops Tween 21 in 30 to 50 mL water.

Reference: Joint Research Centre (2018) Certification Report – The Certification of Particle Size distribution of Corundum: ERM-FD069, Luxembourg: Publications office of the European Union

26



Most Common Statistics





What conclusion can be reached from the data

- The distribution became larger or smaller
- Some aspect of the distribution has changed
- The particle distribution got broader or narrower
- The two modes can be analyzed separately
- There are too many coarse particles in the sample
- The two sets are data are similar or different
- The two techniques produce similar or different data





Which Analyzer?

Size, desired resolution, and budget determine technology and product. For a given problem the choice is often clear.



Want to know more about this particle series? Sign up for the newsletter:

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Modern Particle Characterization Technique Series II: Laser Diffraction

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Thank you



