

# Effortless Emulsion Evaluation for Everyone



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What is an emulsion?

Measuring size->laser diffraction

Measuring size->dynamic light scattering

Measuring charge->zeta potential

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#### "Mixture" of two immiscible liquids such as oil in water.





#### Also need an emulsifier!

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# **Emulsifier?**

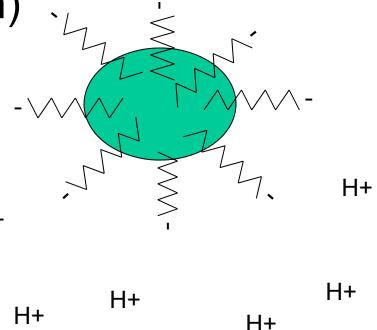


#### Droplets are not stable due to large surface area.

H+

Use an emulsifier to reduce surface energy (surface tension)





# **Examples of Emulsions**



#### Food

- Salad dressings
- Mayonnaise (oil in water, high particle conc.)
- Butter (water in oil)
- Flavoring



- Pharma/Personal Care
  Lotions/Creams
  - Drug delivery



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# Industrial Emulsions

#### Drilling Fluid

 MI SWACO Schlumberger has a reversible oil-water emulsion!

https://www.slb.com/~/media/Files/resources/oilfield\_review/ors04/aut04/06\_reversible\_drilling.pdf

Cutting Fluid

- Water for heat removal/environmental friendliness
- Oil for lubrication

#### Waste

 Waste that includes oil droplets that are emulsified cannot be treated by simple flotation. The emulsion should be broken.



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- You often hear of emulsion polymers. When prepared, they are suspensions of polymer particles in a liquid.
- One step in their preparation is the preparation of an oil (monomer) in water emulsion.





#### Droplet Size

- Affects mouth feel and flavor for food
- Affects kinetics of chemistry (e.g., flavor or drug release)
- Affects suspension viscosity
- Affects suspension appearance
- Affects suspension stability
- Surface Charge
  - Affects suspension stability





What is an emulsion?

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Measuring size->dynamic light scattering

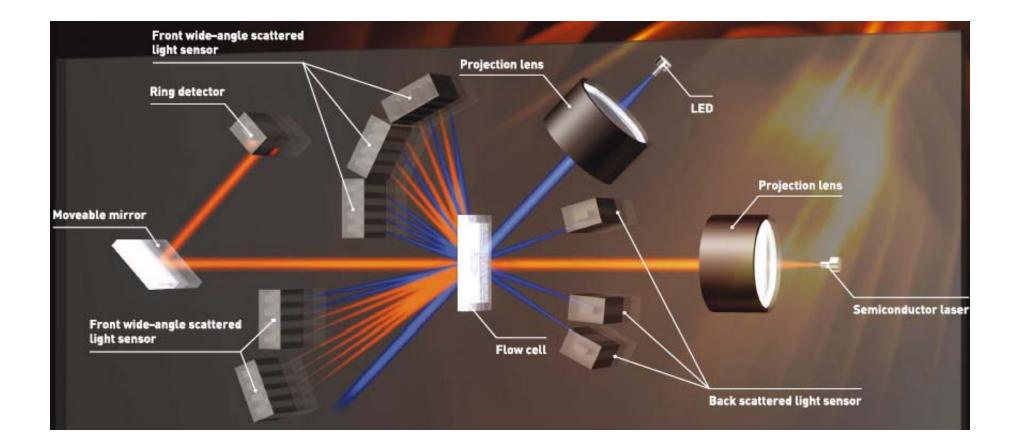
Measuring charge->zeta potential

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#### The Measurement: LA-950 Optics





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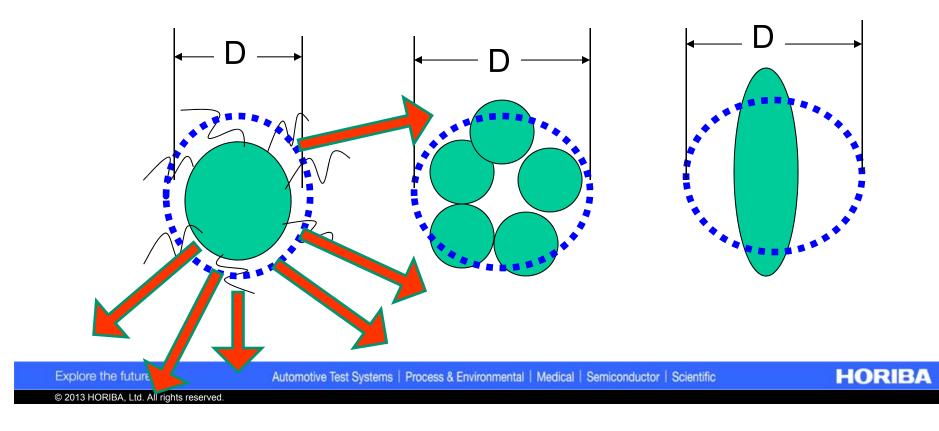
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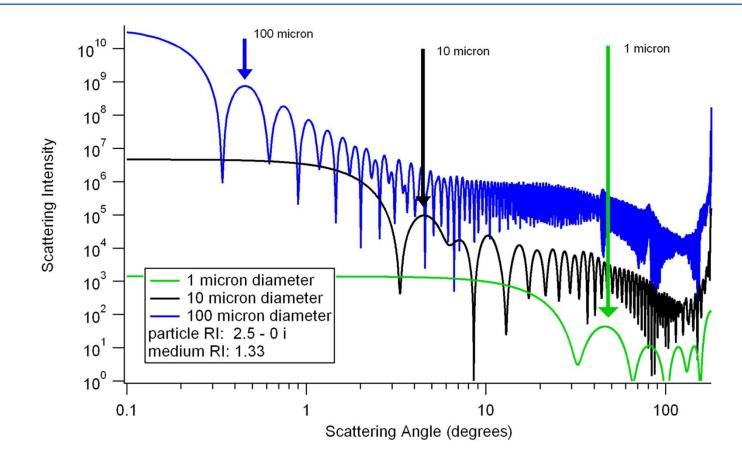
Laser diffraction gives the diameter of a sphere that <u>scatters</u> the same way as your sample.



#### **Mie: Effect of Size**



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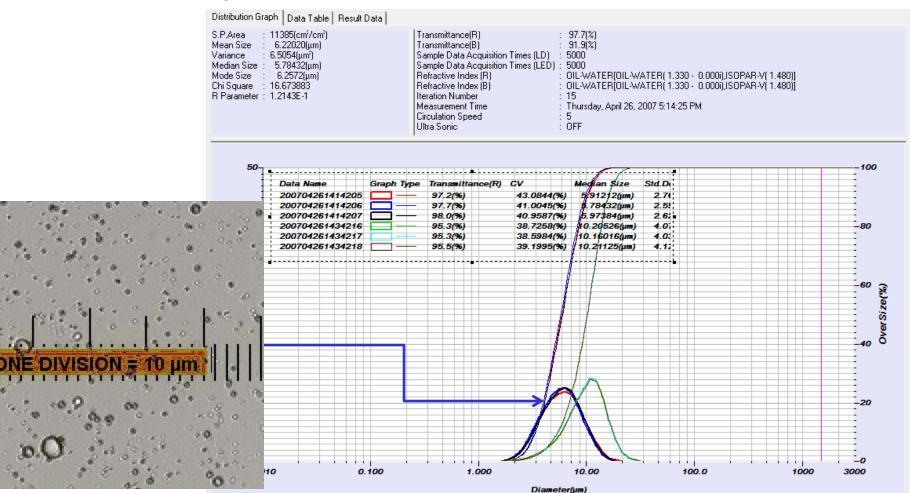
As diameter increases, intensity (per particle) increases and location of first peak shifts to smaller angle.

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# Water in oil emulsion

#### Oil industry



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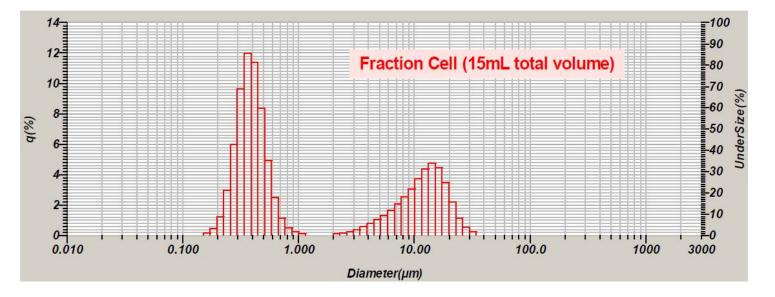
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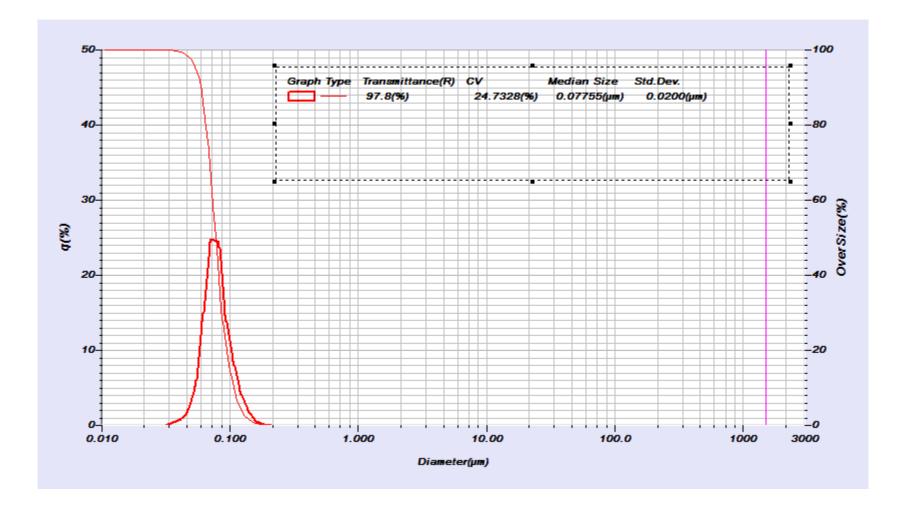
- Bimodal distribution, another water in oil emulsion.
- Used fraction cell for small sample volume.



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### Lotion (oil in water)



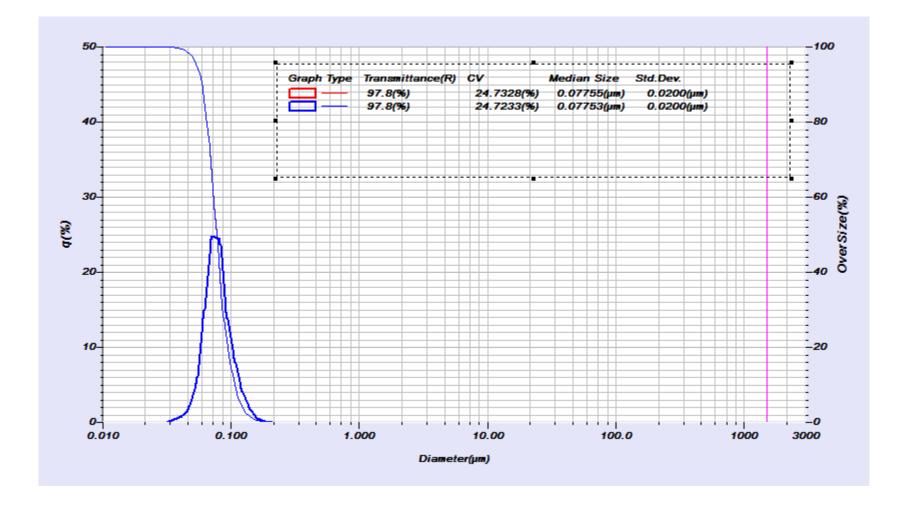


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### Lotion (oil in water)-overlay

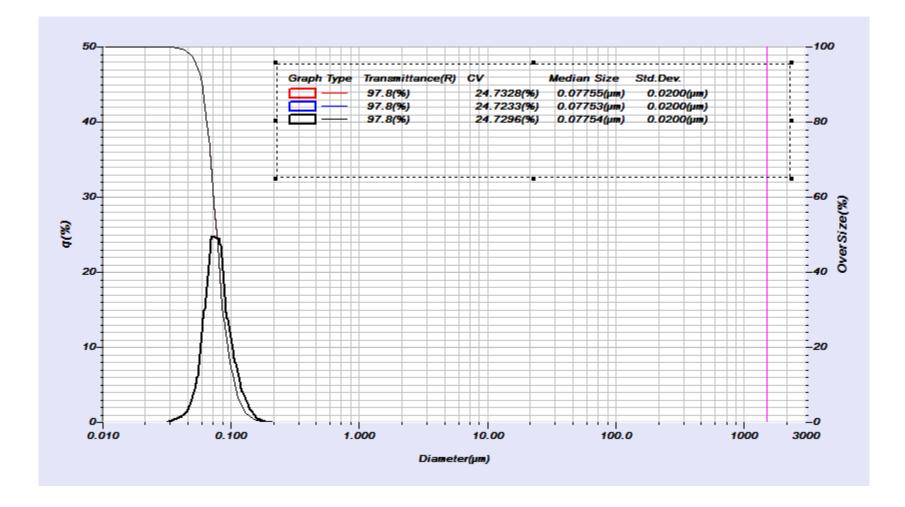


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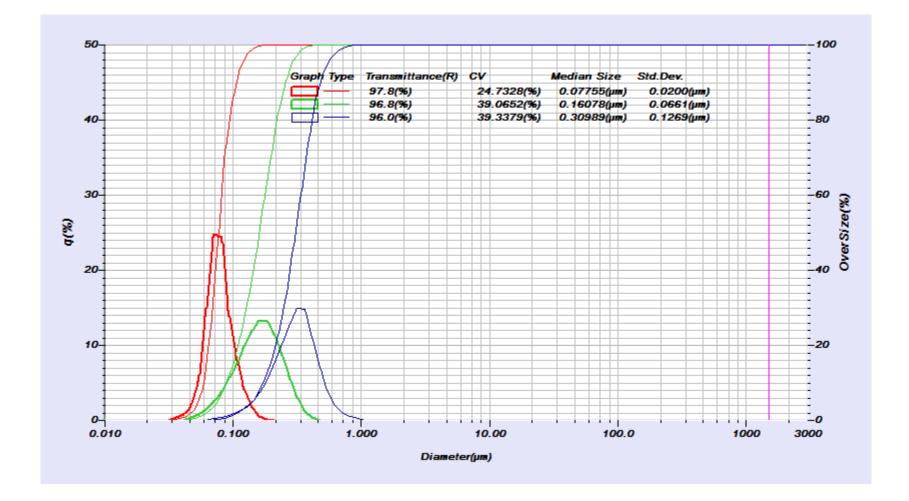


# Lotion (oil in water)-3<sup>rd</sup> repeat



### A variety of formulas









Laser Diffraction can be used to measure:

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Laser Diffraction can be used to measure:

Any emulsion, but check chemical compatibility.

Instruments with chemically resistant parts are easy, but somewhat more expensive to build. Check this point when using (or purchasing) an instrument.





What is an emulsion?

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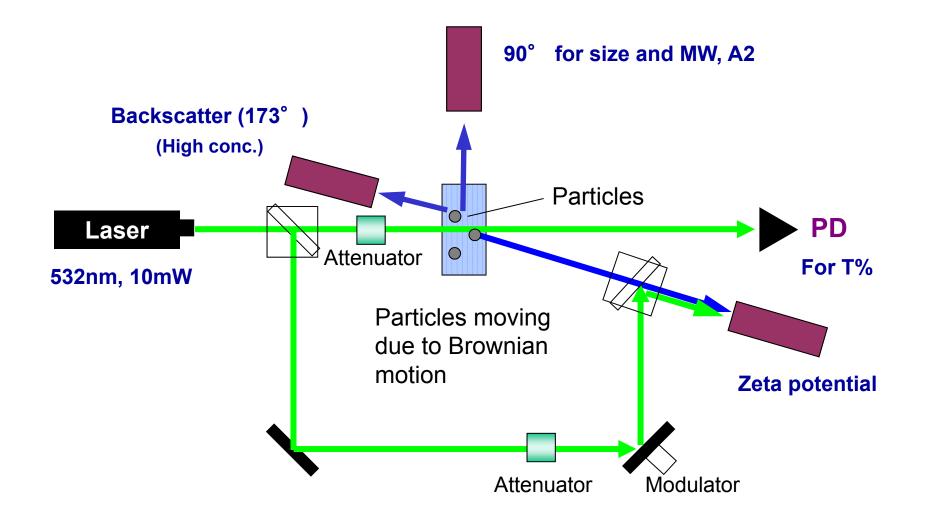
Measuring charge->zeta potential

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# **DLS Optics**



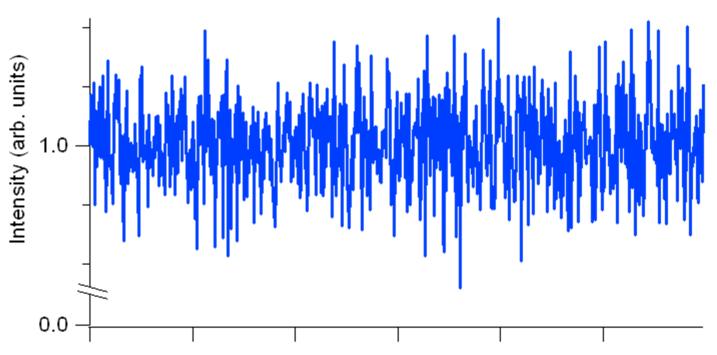


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# **DLS** signal



Random motion of particles leads to random fluctuations in signal (due to changing constructive/destructive interference of scattered light.

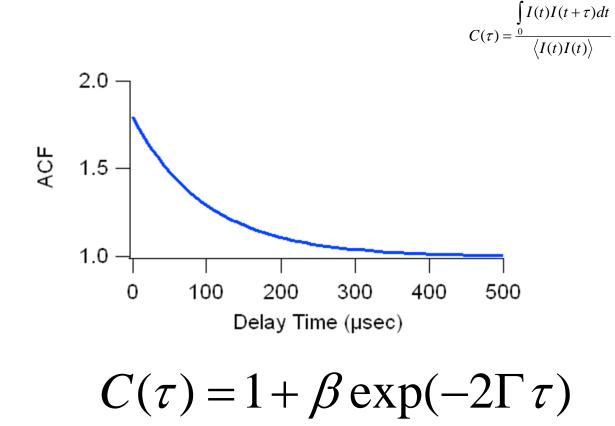


time (microseconds)

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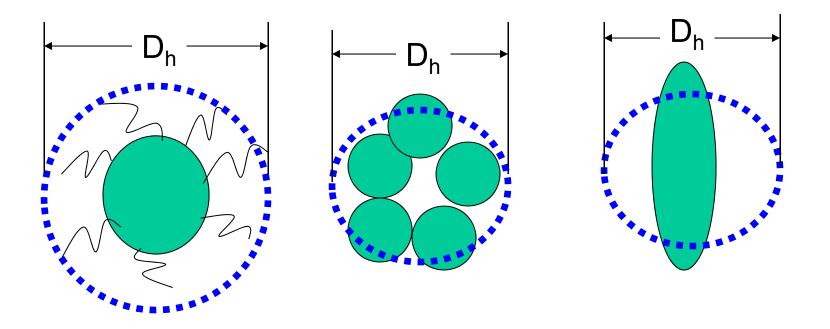


Random fluctuations are interpreted in terms of the autocorrelation function (ACF).





DLS gives the diameter of a sphere that <u>moves</u> (diffuses) the same way as your sample.



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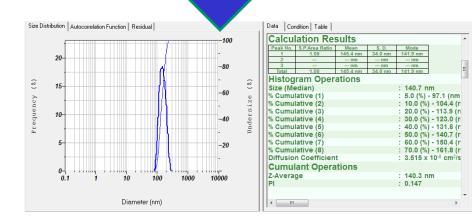


### **Making the Measurement**

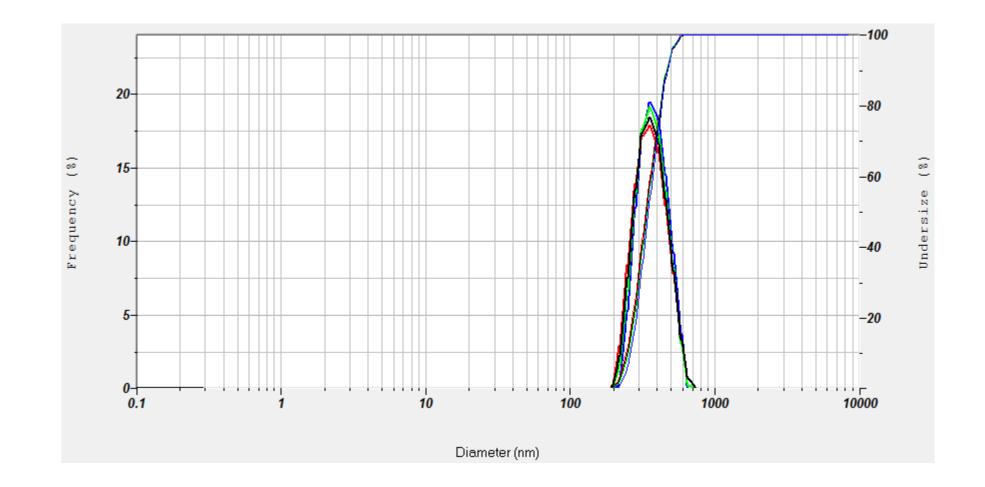




- 1. Fill sample cell.
- 2. Insert cell into instrument.
- 3. Click start.
- 4. Wait ~2 minutes.
- 5. Review results.

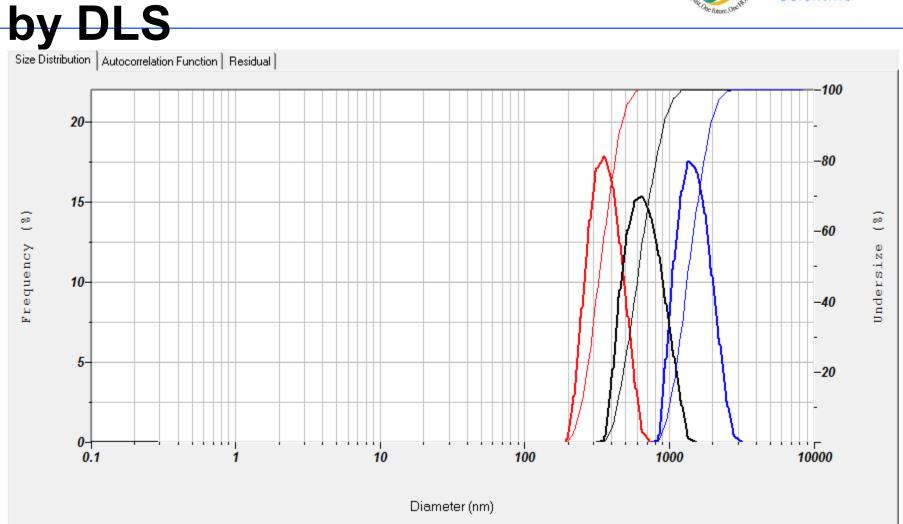






# **Compare 3 flavor emulsions**





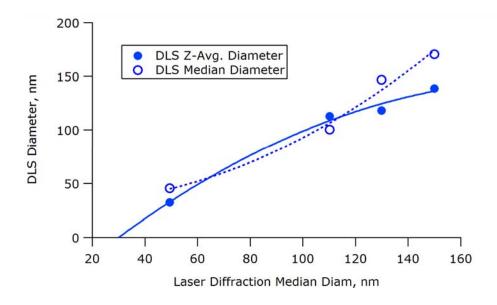
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# **DLS vs Diffraction**



#### Flavor Emulsions, comparing two techniques



Measuring slightly different things so results differ.

#### Which to use?

- What other sizes do you want to measure?
- Stick with Z-average for DLS results
- Choose one technique and stick with it.

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What is an emulsion?

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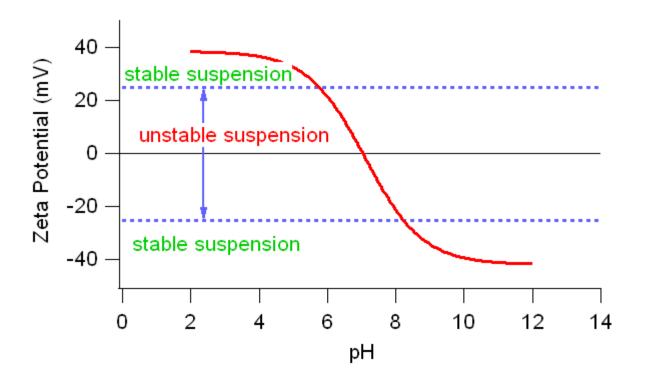
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- Good way of evaluating electrostatic stabilization of suspensions
- Can use to predict interactions

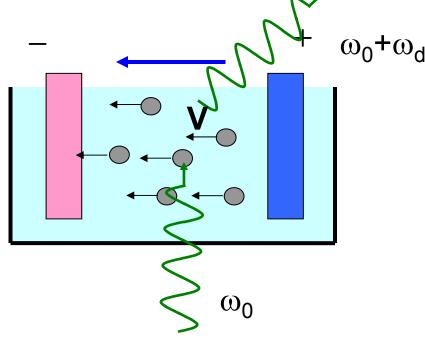


# How to determine zeta



#### potential

- Apply an electric field and probe response of particles to applied field.
- You need to see Doppler shift in scattered light due to particle motion with respect to fixed electrodes.

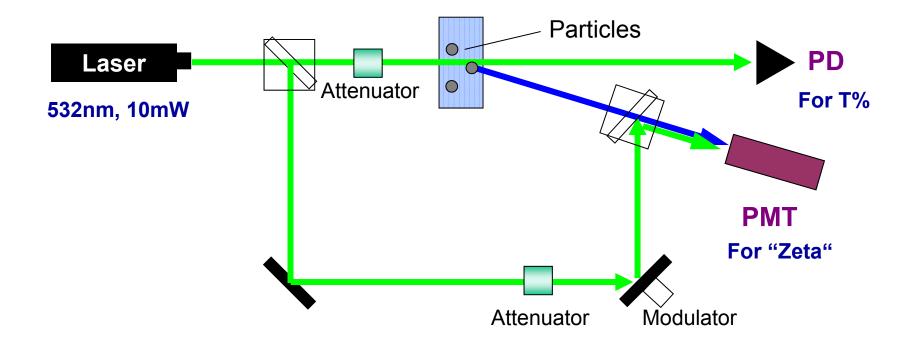








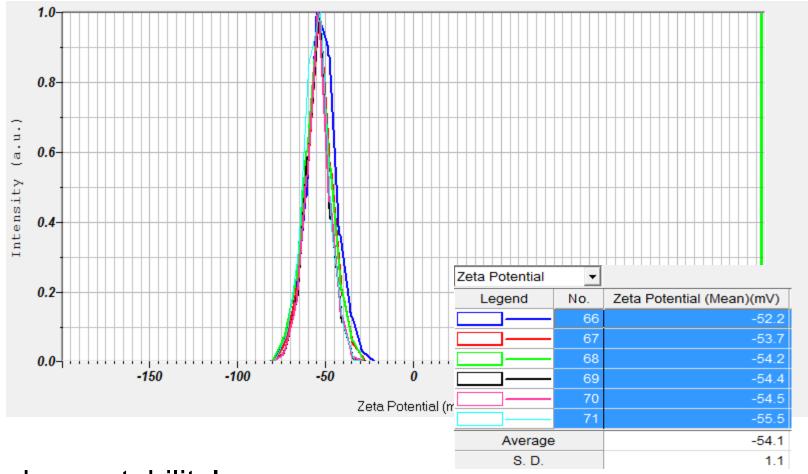
Use optical mixing to extract motion of particles relative to electrodes.



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#### **Zeta Potential of Flavor**





Good repeatability!

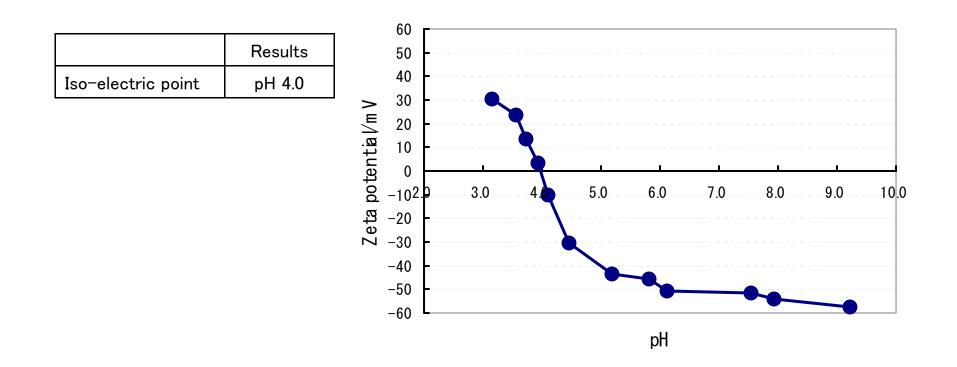
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### **Effect of Liquid**



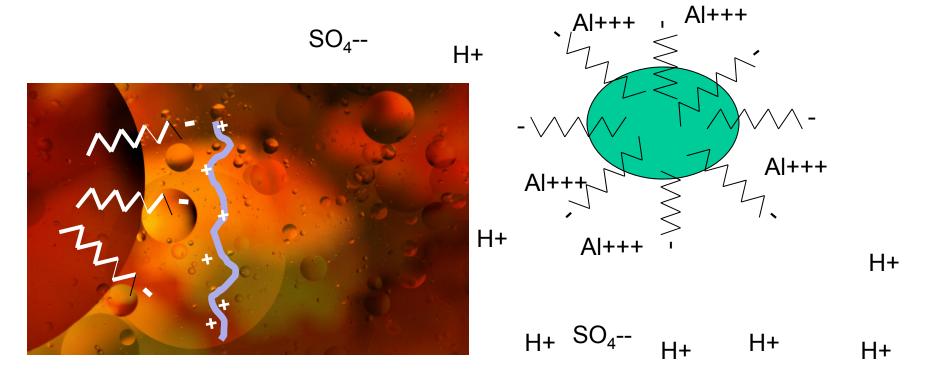
#### ●Iso Electric Point of Coffee Mate



### Coagulent



#### Use polyions to reduce effect of emulsifier and break emulsion or accelerate flocculation.







# When adding coagulant to flocculate waste: ...

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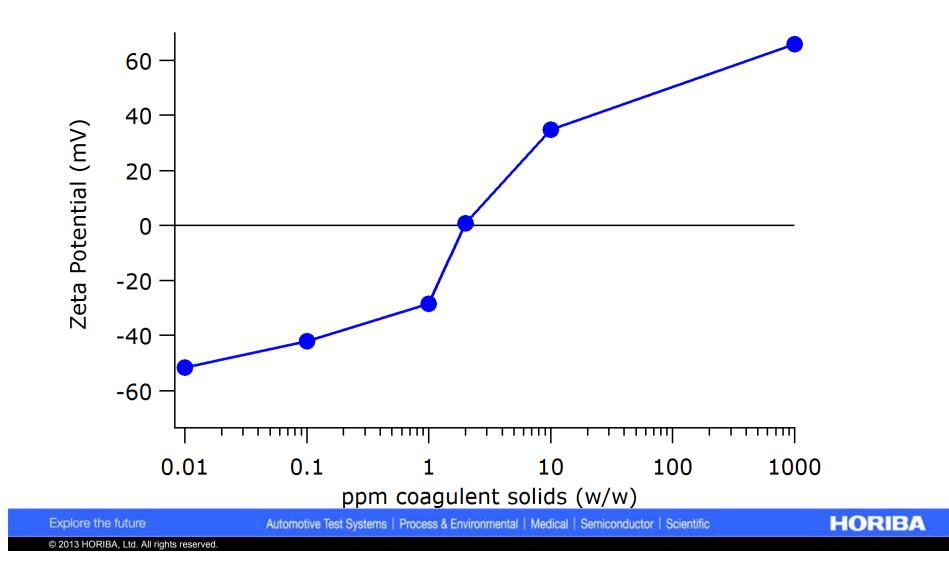
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#### Wastewater



#### Add proprietary coagulent to break emulsion



### Dilution



Sometimes samples need to be diluted. To check if the concentration is correct, do a concentration study.

Dilution DI:sample	Comments	Z-average size, nm (mean of 6 Repeats)
750:1	5 minute	670
	runs	
1000:1	5 minute	657
	runs	
1333:1	5 minute	668
	runs	





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