Research and development support



Battery material evaluation/Causes of deterioration and analysis methods

		HORIBA analytical systems	
Active materials	Cathode	Material distribution evaluation Measurement of carbon concentration in materials Crystallinity evaluation of active materials Observation of active material coating state	Raman Spectroscopy, Laser Scattering Particle Size Distribution Analyze Combustion-infrared absorption method Raman Spectroscopy Raman Spectroscopy, Particle image analysis
		Elution of manganese ions Stoichiometric abnormality in lithium	ICP-OES Glow Discharge Optical Emission Spectrometry Combustion-infrared absorption method. X-ray Fluorescence Analyzer
		Evaluation of carbon crystallinity due to insertion and separation of lithium ions Foreign material contamination Oxyoren deficiency	Raman Spectroscopy X-ray Fluorescence Analyzer Inert das fusion-infrared absorption method
	Anode	Elemental analysis of anode surface (effect of cathode active materials) Elemental analysis of anode depth profile (surface to current collecting foil) Localization and cavities in active materials and foreign material contamination Decrease in reactivity due to SEI formation Evaluation of carbon crystallinity due to insertion and separation of lithium ions Atomization affecting reactivity	X-ray Fluorescence Analyzer Glow Discharge Optical Emission Spectrometry X-ray Fluorescence Analyzer Glow Discharge Optical Emission Spectrometry Raman Spectroscopy Laser Scattering Particle Size Distribution Analyzer
Binder	Binder PVDF Moisture mixed into NMP SBR pH control in the anode slurry adjustment process. Outgas monitoring		Absorption spectroscopy pH meter, quadrupole mass spectrometry
Current collecting foil Cathode : Al Anode : Cu		Surface oxidation, corrosion Surface oxidation, corrosion	Glow Discharge Optical Emission Spectrometry ,Raman Spectroscop Glow Discharge Optical Emission Spectrometry, Raman Spectroscop
Separator Polyolefin, etc.		Metal foreign material X-ray Fluorescence Analyzer Separator damage caused by particle detachment and burrs X-ray Fluorescence Analyzer	
Solid electrolyte		Lithium distribution in solid electrolytes Structural changes in solid electrolytes	Glow Discharge Optical Emission Spectrometry Raman Spectroscopy

Manufacturing, process analysis, control

Measurement of the electrode material mixing flow rate



(1) Measurement results of NMP moisture concentration



(2) Measurement results of outgas components





Battery and electric vehicle development solutions

HORIBA provides a wide range of development support solutions, from evaluation of on-board batteries themselves and vehicle evaluation based on the simulation of actual use of on-board batteries through to battery model development, optimization design, prototyping, and evaluation.



By simulating the vehicle environment using Test in the LoopTM, battery performance such as battery charging and discharging characteristics and thermal management can be evaluated in an environment incorporated into the vehicle system. You can evaluate the charging and discharging behavior of an actual battery under the condition in which it is connected to the powertrain while a running road load is imposed upon it as well as evaluate battery performance by changing the ambient temperature.

CELL



The E-LAB automobile measurement and test center where HORIBA's advanced solutions can be seen, used, and experienced.

Charging and discharging characteristics evaluation

Features
Support for a power range of up to 1 MW
Electrical efficiency: 95% or above
Posponso spood (t00): Loss than 2 ms



Your Partner in Science

Evaluation

Model

development/prototyping

HORIBA MIRA provides support for the development of efficient, high energy



The charging and discharging characteristics of driving batteries and stationary storage batteries can be evaluated. We have a lineup of battery test stations that support everything from driving cells, modules, and packs to large stationary

> Factory Power system



HORIBA Scientific www.horiba.com/scientific

Bulletin: HRE-3699A



Deterioration analysis

Depth profiling of solid electrolyte in all-solid-state batteries Depth elemental profiling of lithium distribution in the solid electrolyte revealed

segregation of lithium on the solid electrolyte surface. With GD-Profiler 2, you can

Analysis of solid electrolyte powder (sulfide-based)

Baman spectroscopy makes it possible to determine differences between the molecular structures and crystallinity of organic and inorganic compounds and to analyze the crystal structure and molecular structure by a non-destructive, non-contact method while observing the materials through an optical microscope. By Raman spectroscopy imaging, the crystalline structure can be visualized to capture the changes in electrolyte due to deterioration.

Observation of ternary cathode(NMC) degradation



Analysis of gases generated during safety testing Lithium-ion batteries require a high level of safety, and there are various kinds of safety tests.

For example, our analyzer is used during the development stage of vehicles to measure any

inflow of CO gases generated from the nail penetration test of lithium ion battery inside the

vehicle, thus contributing to safety evaluation. Our analyzer can also measure in real time the

concentration of CO and CO₂ gases generated during heat tests of batteries under development.

Gas









Ultimate Raman Microscope LabRAM HR Evolution







Elemental image of the anode surface

Left: It was observed that Mn, which is considered to have eluted from the active materials of the cathode due to battery deterioration, had

Right: It was observed that S, which is predicted to have dissolved from the additive components, had unevenly attached to the electrode terminal section and edge section, which are heated to relatively high temperatures.

EMIA-Step The variable heating temperature function makes it possible to perform quantitative analysis by isolating carbon and sulfur by temperature or state in addition to total

Evaluation of battery active material coating

It has become clear that evenly coating electrolyte on the surface of active material particles is effective for improving battery capacity. The resulting coating film is thin and its state differs depending on each particle, so observing such a state is difficult. However, by using Raman spectroscopy and particle analyzing software, Raman measurement for each particle becomes possible. In addition to visualizing each particle's coating state, analysis using the automatic particle recognition function and another type of analysis on the correlation between particle size and shape information can also be conducted.



Powder dispersion

The particle disperser makes powder dispersion on the sample stage easy and helps to improve analysis efficiency. Instant powder dispersion is possible by simple operations not subject to human error. By using the particle disperser in conjunction with Raman spectroscopy or X-ray analysis, analysis of the molecular structure and components of each particle becomes possible.



Optical Emission Spectrometer (GD-OES) 40 60 Time (sec.) GD-Profiler 2 Overall profile from the surface Enlarged profile near the surface to the current collecting foil



Combined evaluation of cathode materials

The dispersion state of a slurry sample before and after mixing were compared by particle size distribution measurement using high concentration cells and Raman spectroscopy. With high concentration cells, changes in overall particle size can be observed, while the behavior of each component can be observed on the um order with Raman spectroscopy



With Raman spectroscopy, the dispersion state of each component can be visualized. Using the particle analyzing software ParticleFinder enables the particle size to be automatically measured per particle. Thus, comparison of figures before and after mixing becomes possible with this method and software.









Comparison by a histogram

Distribution of componen with Raman imaging using the multivariate analysis CLS





Results of measu ent with undilute solution and flow me It was found that there is a difference in particle size distribution depending on

the kneading. By using high concentration cells, the transmittance, which has thus far been adjusted by the sample concentration based on the dilution of solvent, can be adjusted by the optical path length, which allows even a high concentration sample to be measured at the minimum dilution rate.

Partica LA-960V2

Laser Scattering Particle Size Distribution Analyzer

High concentration cell unit

Results of particle size distribution

nt using high concentration cells

CO measurement inside the vehicle

MENU O

during the battery nail penetration test

Color trend graph

< >

This unit enables measuring at the original liquid concentration without diluting the sample. Cell units for various sample viscosities are available.





Battery heat test

NOX SOX CO CO2 O2 CH4

ortable Gas Analyzer

PG-300 Series















from the cathode, and S, which is considered to be a dissolved component of the additive, were also detected on the anode surface.





The effect of the eluted active materials of the cathode is considered to be one of the causes of deterioration of the anode. Based on the obtained overall profile from the surface to the current analysis. Mn. which is considered to have eluted collecting foil and an enlarged profile of elements near the surface, a large quantity of manganese (Mn), an eluted material of the cathode that is a cause of anode surface deterioration, and sulfur (S) estimated to have dissolved from the additive components were detected. On the anode surface, in addition to carbon (C) and lithium (Li), the major components of the solid electrolyte interphase (SEI), a large quantity of Mn that is estimated to have eluted from the cathode and S, which is considered to be an adhesion of the dissolved additive, were detected.



*The data was provided by Professor Akitoshi Havashi of Osaka Prefecture University Analysis of the effect of the eluted active materials of the cathode on the anode Based on the results of X-ray spectrum Mrs





analysis.

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Raman Imaging System

LabRAM Soleil







attached to the entire anode.

Material composition analysis

Measurement of oxygen concentration in a solid sample containing sulfur

It was difficult to determine the quantity of oxygen in oxygen analysis of a solid sample containing sulfur due to the reaction between sulfur in the sample and carbon in the graphite crucible. Using EMGA-930 with a sulfur-removal trap, oxygen in a sulfide can be measured.



Transfer vessel

Our transfer vessel enables measurement without air exposure of samples that react with water, oxygen, and $\ensuremath{\text{CO}_2}$ in the air and then deteriorate without air exposure

Controlling the concentrations of carbon and sulfur in the cathode materials

The carbon/sulfur analyzer EMIA-Expert can control the carbon concentration in the cathode materials at the ppm level. It can also accurately control the sulfur concentration, which affects battery life and capacity degradation.

Carbon/Sulfur Analyzer

Carbon/Sulfur Analyzer

1-frequency induction furnace type

e heating furnace type



Sample weight	Carbon(mass %)
0.3059	0.0200
0.3061	0.0201
0.3055	0.0199
0.3057	0.0198
0.3059	0.0197
Average	0.0199
Standard deviation	0.0002
Coefficient of variation(%)	0.79

Reference) Concentration analysis of carbon in LiCoO₂

Auto sampler

Up to 20 samples can be set automatic setting of the crucible inside the furnace to disposal of the crucible afte measurement





Contamination inspection

Analysis of metal foreign material on separators



Use of an optical image to set the analysis area



Expanded mapping analysis



by Ø10 µm X-ray

With the X-ray Analytical Microscope XGT-9000, vou can detect micro conductive foreign material (metal foreign material) on the separator, which may cause a malfunction due to short circuiting, and carry out an analysis of its elemental composition. Foreign material can be quickly detected by optical image observation, high-speed screening, and highlighting through image processing.

Transfer vessel

Our transfer vessel controls the environment between the glove box and the Instrument, enabling analysis without air exposure.





